



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification:

H01M 8/06, B01J 15/00, 12/00, 19/00,
C01B 3/58, C10K 3/04

A1

(11) International Publication Number:

WO 97/25752

(43) International Publication Date:

17 July 1997 (17.07.97)

(21) International Application Number: PCT/US97/00491

(22) International Filing Date: 10 January 1997 (10.01.97)

(30) Priority Data:

584,512

11 January 1996 (11.01.96)

US

(71) Applicant: INTERNATIONAL FUEL CELLS CORPORATION [US/US]; 195 Governor's Highway, South Windsor, CT 06074 (US).

(72) Inventor: LESIEUR, Roger, R.; 14 Van Buren Road, Enfield, CT 06082 (US).

(74) Agent: KELLY, Robert, H.; Intellectual Property Counsel, Hamilton Standard, Patent Dept. MS 1-2-AB24, One Hamilton Road, Windsor Locks, CT 06096-1010 (US).

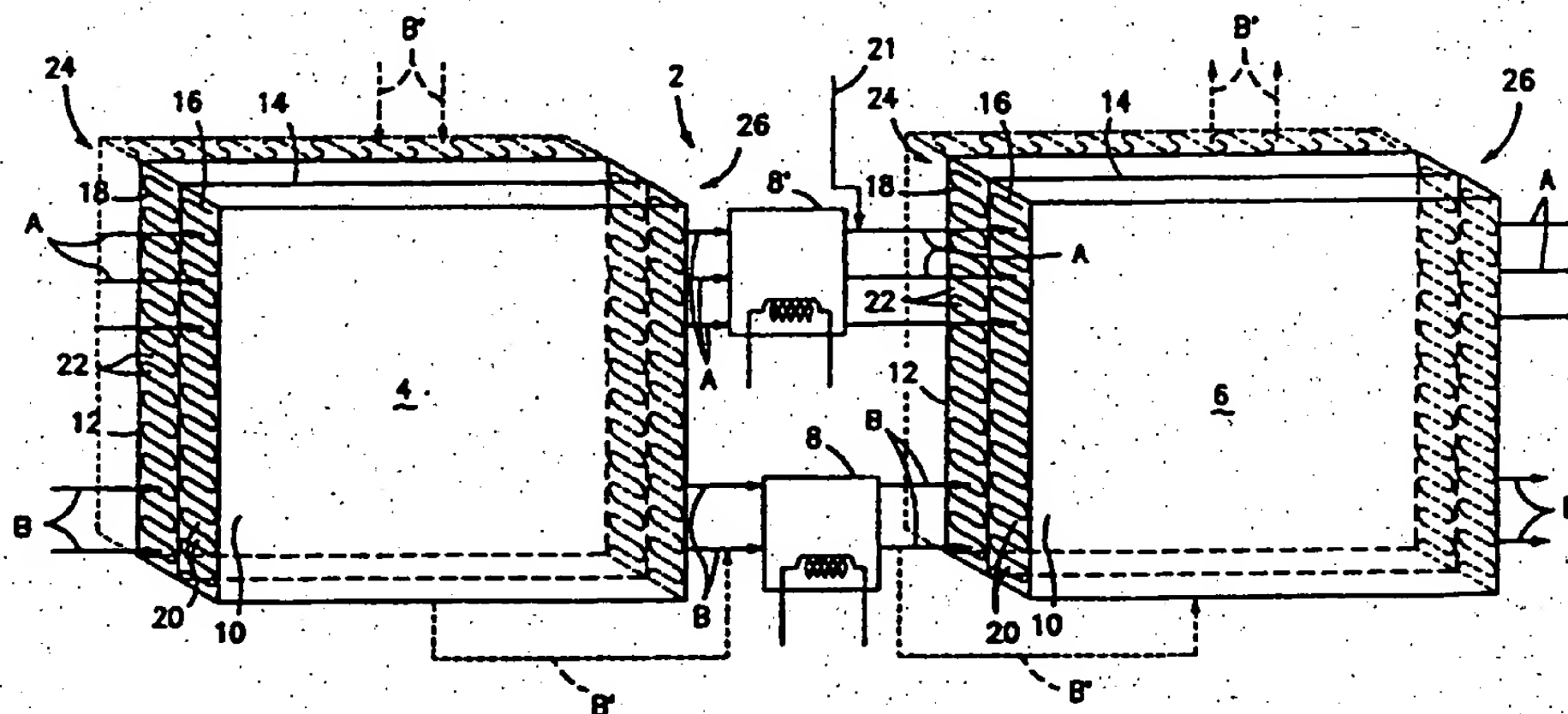
(81) Designated States: CA, JP, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).

Published

With international search report.

Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.

(54) Title: COMPACT SELECTIVE OXIDIZER ASSEMBLAGE FOR FUEL CELL POWER PLANT



(57) Abstract

A fuel gas selective oxidizer assemblage (2) for use in a fuel cell power plant is formed from a series of repeating components, each of which includes a plurality of separate selective oxidizer gas passages (20) and a plurality of adjacent coolant passages (22). The selective oxidizer gas and coolant passages are preferably formed by corrugated metal sheets (16, 18) which are interposed between planar metal sheets (10, 12, 14). One of the planar metal sheets (14) forms a common wall between the selective oxidizer gas passages and the coolant passages so as to provide enhanced heat transfer between the reformat gas in the selective oxidizer gas passages and the coolant. Each of the components includes a first outer planar metal sheet (10), a medial planar sheet (14), and a second outer planar metal sheet (12). Each of the planar metal sheets is separated from the next by the corrugated metal gas (16) and coolant passage sheets (18).

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AM	Armenia	GB	United Kingdom	MW	Malawi
AT	Austria	GE	Georgia	MX	Mexico
AU	Australia	GN	Guinea	NE	Niger
BB	Barbados	GR	Greece	NL	Netherlands
BE	Belgium	HU	Hungary	NO	Norway
BF	Burkina Faso	IE	Ireland	NZ	New Zealand
BG	Bulgaria	IT	Italy	PL	Poland
BJ	Benin	JP	Japan	PT	Portugal
BR	Brazil	KE	Kenya	RO	Romania
BY	Belarus	KG	Kyrgyzstan	RU	Russian Federation
CA	Canada	KP	Democratic People's Republic of Korea	SD	Sudan
CF	Central African Republic	KR	Republic of Korea	SE	Sweden
CG	Congo	KZ	Kazakhstan	SG	Singapore
CH	Switzerland	LI	Liechtenstein	SI	Slovenia
CI	Côte d'Ivoire	LK	Sri Lanka	SK	Slovakia
CM	Cameroon	LR	Liberia	SN	Senegal
CN	China	LT	Lithuania	SZ	Swaziland
CS	Czechoslovakia	LU	Luxembourg	TD	Chad
CZ	Czech Republic	LV	Latvia	TG	Togo
DE	Germany	MC	Monaco	TJ	Tajikistan
DK	Denmark	MD	Republic of Moldova	TT	Trinidad and Tobago
EE	Estonia	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	UG	Uganda
FI	Finland	MN	Mongolia	US	United States of America
FR	France	MR	Mauritania	UZ	Uzbekistan
GA	Gabon			VN	Viet Nam

Description

Compact Selective Oxidizer Assemblage for Fuel Cell Power Plant

Technical Field

5 This invention relates to a selective oxidizer assemblage which is formed from a plurality of repeating sub-assemblies. More particularly, this invention relates to a fuel gas selective oxidizer assemblage which is compact and lighter in weight than conventional oxidizer assemblages used in fuel cell power plants.

Background Art

10 Fuel cell power plants include fuel gas selective oxidizers which are operable to remove carbon monoxide from a reformed fuel gas, such as natural gas, which is used as a fuel for fuel cell power plants. The procedure involves passing the reformed fuel gas with small amounts of added gaseous oxygen through a catalytic bed which is capable of
15 oxidizing carbon monoxide in an exothermic reaction. The reaction proceeds at controlled temperatures which are within a given range of about 360o F. to about 170o F.. The temperature of the catalyst bed must be maintained above a particular threshold temperature which is between about 220o F. to about 360o F. at the entry stage of the catalyst bed, where
20 the gases being treated are relatively rich in carbon monoxide, and will be reduced to lower temperatures of about 170o F.-220o F. at latter stages of the catalyst bed where the carbon monoxide content of the gas is lower. The catalysts typically used are platinum catalysts which are deposited on alumina granules. U.S. Patent No. 5,330,727, granted July 19, 1994 to J. C.
25 Trocciola et al discloses a selective oxidizer assemblage which is proposed for use in a fuel cell power plant and describes the temperature regimes

for use in a fuel cell power plant and describes the temperature regimes required to properly oxidize the carbon monoxide. The type of oxidizer shown in the aforesaid patent is conventionally referred to as a "shell and tube" heat exchanger.

5 The shell and tube fuel cell power plant selective oxidizers require a large amount of heat transfer surface area between the catalyst bed and the coolant in order to maintain the controlled temperatures needed to produce the degree of carbon monoxide oxidization required to operate the fuel cells properly. This need for large heat transfer surface area, when met by using
10 catalyst-coated granules requires that the catalyst coated granules be diluted, which results in undesirably large and heavy oxidizer assemblies. For example, a 20 KW acid fuel cell power plant that includes a shell and tube oxidizer component requires a volume of about 4 cubic feet for the oxidizer. Higher power fuel cell power plants, such as 200 KW plants or
15 larger, will require proportionally larger fuel gas oxidizers.

It would be highly desirable to provide a fuel oxidizer which is suitable for use in a fuel cell power plant, which oxidizer supplies the necessary catalyzed and coolant surface area, but is compact, strong, and light in weight.

20 Disclosure of the Invention

This invention relates to a selective oxidizer structure which provides the necessary catalyzed surface area, and heat transfer surface area, is substantially smaller and lighter than presently available tube and shell selective oxidizers, and can provide enhanced temperature control
25 throughout the length of the device. The selective oxidizer structure of this invention is formed from a series of essentially flat plate heat exchanger components. Each of the heat exchanger components includes reformed gas passages and adjacent heat exchanger coolant passages. At the entry end of the oxidizer assembly, the reformed fuel gas passages are

connected to a fuel gas line which feeds the fuel gas mixture into the oxidizer from the reformer and shift converter assemblies. The opposite exit end of the oxidizer assembly connects to a line that directs the treated fuel gas mixture emanating from the oxidizer assembly to the fuel cell stack in the power plant. The direction of coolant flow can be the same as, counter to, or perpendicular to, the direction of flow of the fuel gas through the device.

The flat plate components of the selective oxidizer assembly may be formed from flat metal sheets which are separated from each other by corrugated metal sheets, or by U-shaped strips, as will be described hereinafter. The corrugated sheets provide the high catalyzed surface area on the gas passage side of the device needed to properly oxidize the carbon monoxide constituent in the fuel gas. The corrugated sheets also provide an extended heat transfer surface for the device. The metal sheets which make up the fuel gas passage components have all of their fuel gas-contacting surfaces coated with a catalyzed alumina layer that is applied to the gas-contacting surfaces by means of a conventional process such as provided by W.R. Grace and Co. The process is presently used to produce automobile catalytic converters, wood stove catalytic emission units, and the like. The metal plates used to form the flat plate components are preferably steel alloy plates containing aluminum which can be brazed or spot welded together; surface oxidized; primed with a wash coat; and finally coated with a catalyst which, when dried, adheres to the wash coated surfaces of the plates. As noted above, only the fuel gas passages in the assembly are catalyzed. The use of a series of separate passages in each fuel gas flow section in the oxidizer provides the necessary catalyzed surface area. Heat transfer from the stream of fuel gas being oxidized is more readily controlled by essentially pairing each gas passage in the device with its own coolant passage, with the paired gas and coolant passages sharing a common wall. Thus, the use of the flat plate construction enables more accurate control of the operating temperatures of the oxidizer. The benefits of the sandwiched

plate construction of this invention are possible because this construction can provide up to five hundred square feet of heat exchange surface per cubic foot of volume with the catalyst in intimate contact with the heat exchange surface of the device. Using the design of this invention, a selective oxidizer for use with a 20 KW power plant can be formed using only 0.1 cubic foot of space.

It is therefore an object of this invention to provide an improved selective oxidizer assembly for use in a fuel cell power plant, which oxidizer assembly is compact and lightweight.

It is a further object of this invention to provide an oxidizer assembly of the character described which operates at more accurately controlled temperatures than presently available oxidizer assemblies.

It is another object of this invention to provide an oxidizer assembly of the character described which is inexpensive to manufacture as compared to commercially available fuel cell power plant fuel gas selective oxidizers.

These and other objects and advantages of this invention will become readily apparent to one skilled in the art from the following detailed description of a preferred embodiment of the invention when taken in conjunction with the accompanying drawings in which:

Brief Description of the Drawings

FIG. 1 is a schematic view of a first embodiment of a selective oxidizer assembly formed in accordance with this invention;

FIG. 2 is a schematic view similar to FIG. 1 of a second embodiment of a selective oxidizer assembly formed in accordance with this invention; and

FIG. 3 is an end view of a selective oxidizer assembly formed in accordance with this invention showing the manner of construction of the gas and coolant passages.

Best Mode For Carrying Out This Invention

Referring now to the drawings, there is shown in FIG. 1 a schematic view of a first embodiment of a fuel gas selective oxidizer assembly formed in accordance with this invention. The selective oxidizer assembly is denoted generally by the numeral 2 and includes a pair of spaced-apart heat exchange zones 4 and 6, with a coolant heat exchanger 8 interposed between the two zones 4 and 6. The oxidizer assembly 2 is formed from parallel outer planar wall members 10 and 12 sandwiched around, and spaced-apart from a medial planar wall member 14. the wall members 10, 12 and 14 are separated from each other by corrugated core sheets 16 and 18. The planar sheets 10 and 14 combine with the corrugated sheet 16 to form a plurality of gas flow passages 20 through which the reformer gas-oxygen mixture flows. The planar sheets 12 and 14 combine with the corrugated sheet 18 to form a plurality of coolant flow passages 22.

It will be noted that each of the gas flow passages 20 is paired with a respective coolant flow passage 22, and that the passages 20 and 22 share a common wall, i.e., the planar sheet 14. The embodiment shown in FIG. 1 is an embodiment of the invention which employs unidirectional flow of both the coolant stream and the reformer gas-oxygen mixture stream. That is to say, the fuel gas-oxygen mixture stream flows in the direction indicated by arrows A, and the coolant stream flows in the same direction, as indicated by arrows B. The ends of the assembly 2 denoted by the numeral 24 can be characterized as inlet ends for the assembly zones 4 and 6; and the ends of zones 4 and 6 of the assembly denoted by the numeral 26 can be characterized as outlet ends of the assembly. It will be understood that the gas mixture entering the passages 20 at the inlet ends 24 of the zones 4 and 6 of the assembly 2 comes from the fuel gas reformer and shift converter components of the power plant, and the gas mixture leaving the

passages 20 from the zone 6 is piped to the active area of the power plant cell stack.

The assembly 2 operates as follows. Referring briefly to FIG. 3, the walls 11 and 15 of the plates 10 and 14 respectively, as well as the surfaces of the corrugated sheet 16 which form the sides of the gas passages 20 in the zones 4 and 6, are provided with a platinum catalyst coating 21 which is capable of selectively oxidizing carbon monoxide (CO) in the reformer fuel gas-oxygen mixture. The fuel gas-oxygen mixture, which may contain as much as 1.0% (10,000 ppm) CO enters the inlet end 24 (shown in FIG. 1) of the fuel passages 20 at temperatures which are typically in the range of about 220o F. to about 360o F.. The coolant stream enters the inlet end 24 (shown in FIG. 1) of the coolant passages 22 at temperatures in the range of about 180o F. to about 360o F.. It is important to maintain the fuel gas mixture at a temperature of not less than about 220o F. at the inlet end 24 of the assemblage 2 so as to ensure that the catalyst in the fuel gas mixture flow passages 20 is not rendered ineffective by the relatively large percentage of CO in the gas mixture as it enters the oxidizer zone 4. The threshold temperature for catalyst degradation is 220o F., however if the temperature increases to greater than about 360o F. to about 380o F., the catalyst loses its selectivity, and the hydrogen in the reformed gas stream will be burned, in lieu of the CO in the gas stream. The high heat transfer provided by this assemblage with its increased catalyzed surfaces, and the heat transfer fluid (either gas or liquid), maintains the reactant stream at an optimum temperature, even when there is a large quantity of heat generated from oxidation of CO in the gas stream. The coolant and gas streams both exit the first zone 4 of the assemblage at a temperature in the range of about 220o F. to about 380o F..

At this point, the CO content of the reformer gas stream is typically in the range of about 300-500 ppm. The coolant stream passes through the coolant heat exchanger 8 where the temperature of the coolant stream is lowered to about 170o F.. The 170o F. coolant stream enters the coolant

passages 22 at the entrance end 24 of the second zone 6 of the
assemblage 2, while at the same time the fuel gas stream is cooled, if
necessary, to a temperature in the range of about 170o F. to about 220o F.
in a fuel gas heat exchanger 8'. Additional oxygen may be added to the fuel
5 gas line via line 21. The re-oxygenated fuel gas then enters the gas stream
passages 22 at the entrance end 24 of the second zone 6 of the
assemblage 2. Flowing through the second zone 6 of the assemblage 2,
the temperature of the reformer gas stream will be lowered to about 180o F.,
and the CO content of the reformer gas stream will be lowered to less than
10 about 10 ppm. The reformer gas stream which exits the end 26 of the
assemblage zone 6 will thus have a temperature of about 180o F. and a CO
content of less than about 10 ppm. As shown in phantom lines in FIG. 1,
the coolant and fuel gas streams can be made to flow in cross directions as
noted by the arrows A and B'.

15 FIG. 2 is a schematic view of a second embodiment of a selective
oxidizer which is formed in accordance with this invention. The assemblage
shown in FIG. 2 is denoted generally by the numeral 2' and includes a
housing 28 formed from the flat plate components as described above. The
coolant stream flows through the housing 28 in the direction of the arrows D
20 and the fuel gas-oxygen mixture stream flows through the housing 28 in the
direction of the arrows C. The coolant enters the end 30 of the housing 28
at a temperature which is less than about 220o F., and preferably in the
range of about 150o F. to about 170o F.. The fuel gas-oxygen mixture
stream enters the end 32 of the housing 28 at a temperature in the range of
25 about 220o F. to about 380o F.. The fuel gas stream exits the end 30 of the
housing 28 at a temperature in the range of about 180o F. to about 220o F.,
and the coolant stream exits the end 32 of the housing 28 at a temperature
in the range of about 220o F. to about 360o F..

30 The use of the plate construction with outer planar parts and inner
separate passages results in a lightweight, strong oxidizer assembly which
provides a large surface area per unit volume. The entire surface of the fuel

gas flow passages in the oxidizer heat exchanger assembly can be catalyzed by wash coating and selectively applying the catalyst to the fuel gas flow passages in the assembled structure. The fact that the gas flow and coolant flow sections of the assembly form extensive heat exchange surfaces allows the assembly to be operated at more accurately controlled temperatures than the currently available shell and tube catalyzed pellet - type oxidizers. The gas and coolant passages have been shown in the drawings as being formed from one or more corrugated sheets, however, separate U-shaped strips could also be used instead. The weight and size savings achieved by using the plate-type construction described above is enhanced with larger higher power output fuel cell power plants.

Since many changes and variations of the disclosed embodiment of the invention may be made without departing from the inventive concept, it is not intended to limit the invention other than as required by the appended claims.

What is claimed is:

Claims

1. A hydrocarbon fuel gas selective oxidizer assembly comprising:

a) a first inlet line for directing a mixture of the fuel gas and oxygen into the assembly;

b) a gas flow section connected to said first inlet line so as to receive a stream of the fuel gas and oxygen mixture, said gas flow section being formed from first and second spaced-apart plates with a plurality of separate gas passages sandwiched between said first and second plates;

c) a second inlet line for directing a coolant into the assembly; d) a coolant section connected to said second inlet, said coolant section being adjacent to said gas flow section, said coolant section being formed from said first plate and a third plate which is spaced-apart from said first plate on a side thereof opposite said second plate, said coolant section further including a plurality of coolant passages sandwiched between said first and third plates, said first plate providing heat transfer from gas flowing through said gas passages to coolant flowing through said coolant passages;

e) a first outlet line connected to said gas flow section for removing oxidized fuel gas from the assembly; and

f) a second outlet line connected to said coolant section for removing coolant from the assembly.

2. The selective oxidizer assembly of Claim 1, wherein walls of said gas passages are covered with a platinum catalyst coating which is operable to selectively oxidize carbon monoxide in fuel gas flowing through said gas passages.

3. The selective oxidizer assembly of Claim 1, wherein said first and second inlet lines are disposed on a common side of said assembly so that the fuel gas and oxygen mixture and the coolant flow through the assembly in the same direction.

4. The selective oxidizer assembly of Claim 3, wherein said assembly includes a plurality of adjacent zones, each of said zones including a gas flow section and a coolant section, and said zones being separated by a heat exchanger which receives coolant from said first zone and is operable to lower the temperature of the coolant before the coolant enters said second zone, and wherein said inlet lines are connected to said first zone, and wherein said outlet lines are connected to said second zone.

5. The selective oxidizer assembly of Claim 1, wherein said first and second inlet lines are disposed on opposite sides of said assembly so that the fuel gas and oxygen mixture and the coolant flow through the assembly in opposite directions.

6. The selective oxidizer assembly of Claim 1, wherein said first and second inlet lines are disposed on adjacent sides of said assembly so that the fuel gas and oxygen mixture, and the coolant both flow through the assembly in cross directions.

7. The selective oxidizer assembly of Claim 6, wherein said assembly includes a plurality of adjacent zones, each of said zones including a gas flow section and a coolant section, and said zones being separated by a heat exchanger which receives coolant from said first zone and is operable to lower the temperature of the coolant

before the coolant enters said second zone, and wherein said inlet lines are connected to said first zone, and wherein said outlet lines are connected to said second zone.

8. A hydrocarbon fuel gas selective oxidizer assembly comprising:

a) a first inlet line for directing a mixture of the fuel gas and oxygen into the assembly;

5 b) a gas flow section connected to said first inlet line so as to receive a stream of the fuel gas and oxygen mixture, said gas flow section being formed from first and second spaced-apart plates with a plurality of separate gas passages sandwiched between said first and second plates;

c) a second inlet line for directing a coolant into the assembly;

10 d) a coolant section connected to said second inlet, said coolant section being adjacent to said gas flow section, said coolant section being formed from said first plate and a third plate which is spaced-apart from said first plate on a side thereof opposite said second plate, said coolant section further including a plurality of coolant passages sandwiched between said first and third plates, said first plate providing heat transfer from gas flowing through said gas passages to coolant flowing through said coolant passages;

15 e) a first outlet line connected to said gas flow section for removing oxidized fuel gas from the assembly;

20 f) a second outlet line connected to said coolant section for removing coolant from the assembly; and

g) said first and second inlet and outlet lines being disposed on opposite sides of said assembly so that the fuel gas and oxygen mixture and the coolant flow through the assembly in opposite directions.

25

9. A hydrocarbon fuel gas selective oxidizer assembly comprising:
- a) a first inlet line for directing a mixture of the fuel gas and oxygen into the assembly;
 - b) a gas flow section connected to said first inlet line so as to receive a stream of the fuel gas and oxygen mixture, said gas flow section being formed from first and second spaced-apart plates with a plurality of separate gas passages sandwiched between said first and second plates;
 - c) a second inlet line for directing a coolant into the assembly;
 - d) a coolant section connected to said second inlet, said coolant section being adjacent to said gas flow section, said coolant section being formed from said first plate and a third plate which is spaced-apart from said first plate on a side thereof opposite said second plate, said coolant section further including a plurality of coolant passages sandwiched between said first and third plates, said first plate providing heat transfer from gas flowing through said gas passages to coolant flowing through said coolant passages;
 - e) a first outlet line connected to said gas flow section for removing oxidized fuel gas from the assembly;
 - f) a second outlet line connected to said coolant section for removing coolant from the assembly; and
 - g) said first and second inlet lines being disposed on adjacent sides of said assembly so that the fuel gas and oxygen mixture, and the coolant both flow through the assembly in cross directions.

10. A method for selectively oxidizing reformed fuel gas to convert carbon monoxide in the fuel gas to carbon dioxide, said method comprising the steps of:
- a) providing a stream of a mixture of the fuel gas and oxygen;

5

b) directing said stream of fuel gas and oxygen over an oxidizing platinum catalyst which is operable to selectively oxidize carbon monoxide in said fuel gas stream; and

10

c) providing a coolant stream adjacent to, and in heat exchange relationship with the fuel gas and oxygen stream, said fuel gas stream and said coolant stream flowing in opposite directions to each other.

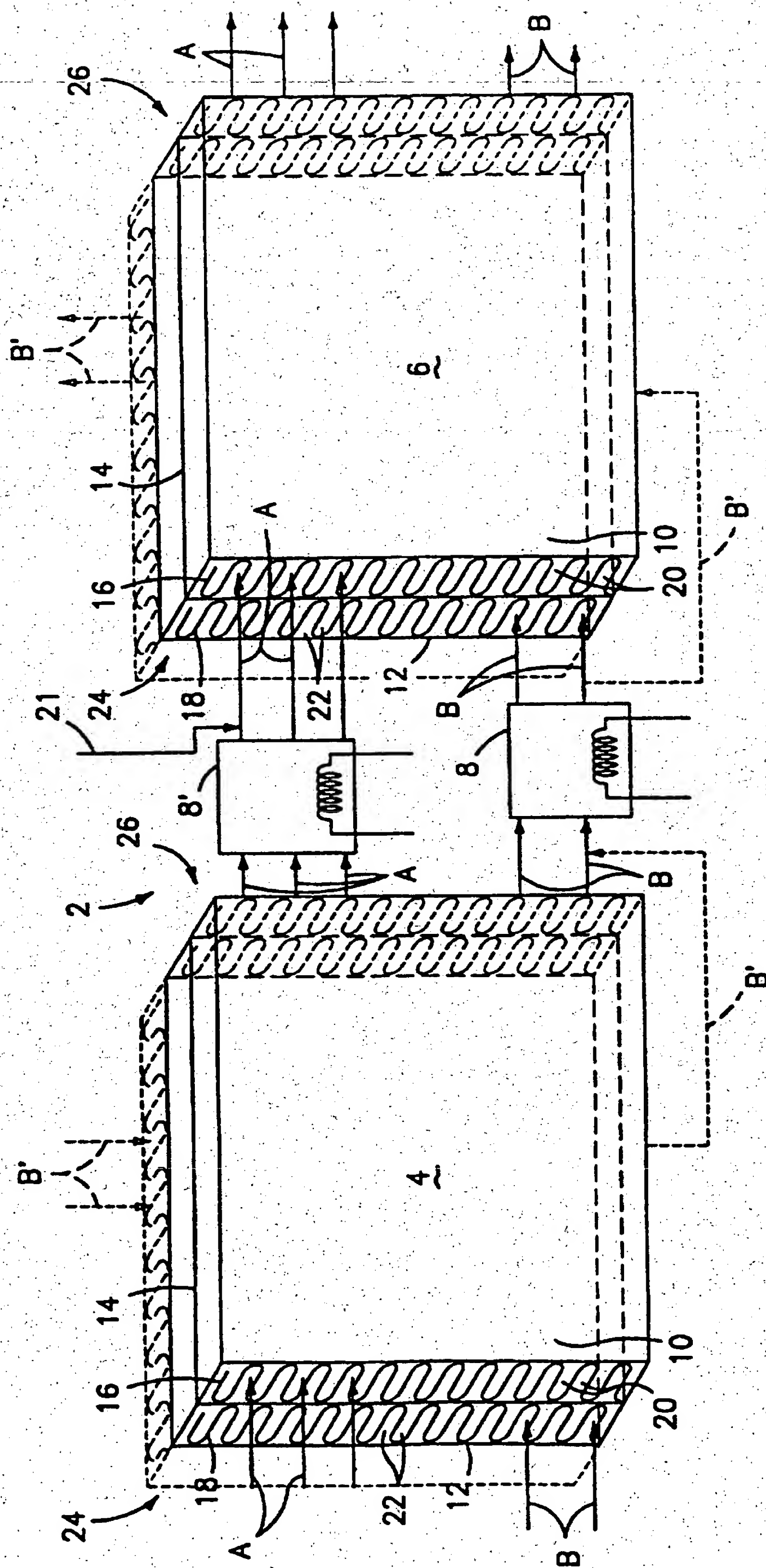


FIG. 1

2/2

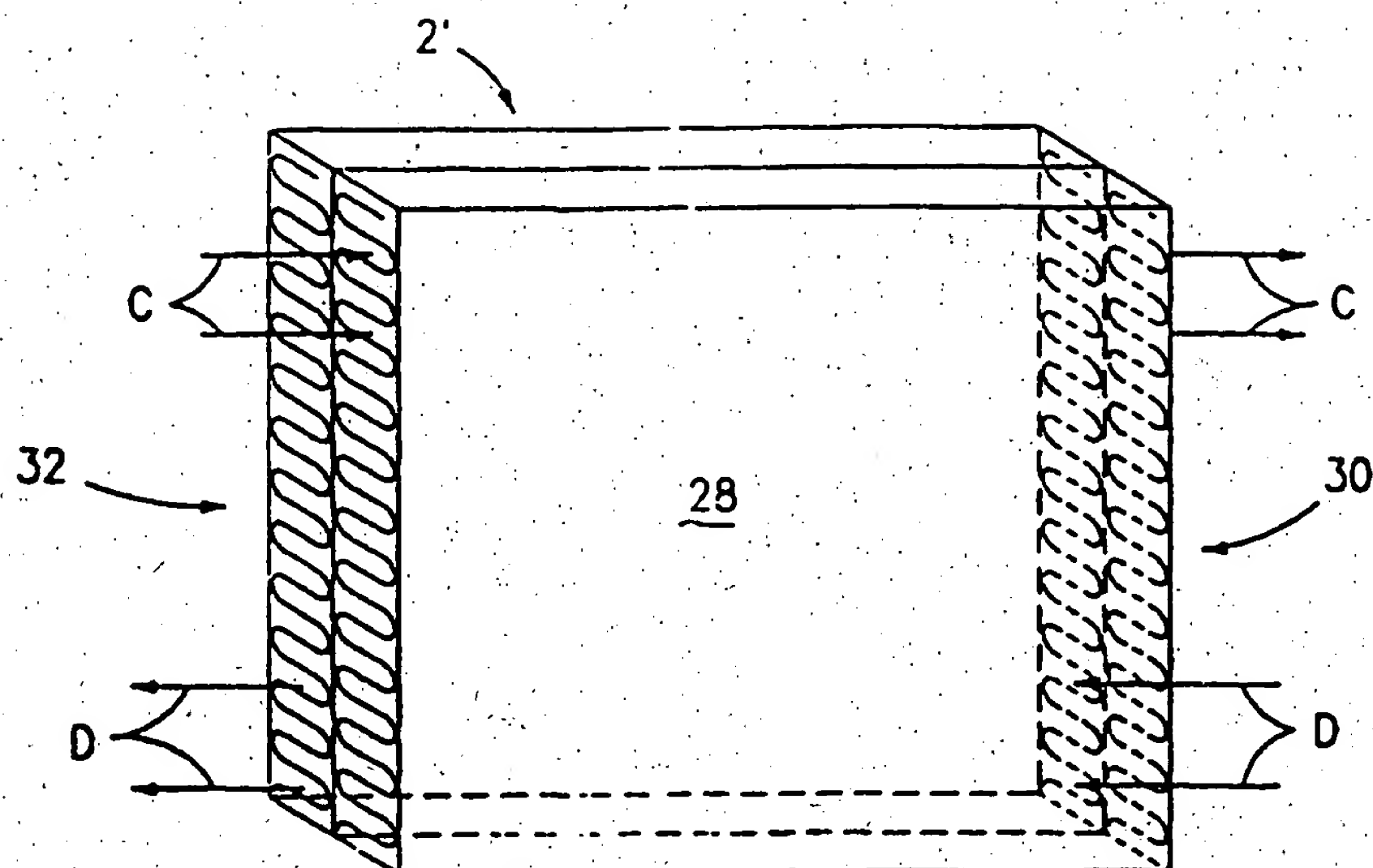


FIG. 2

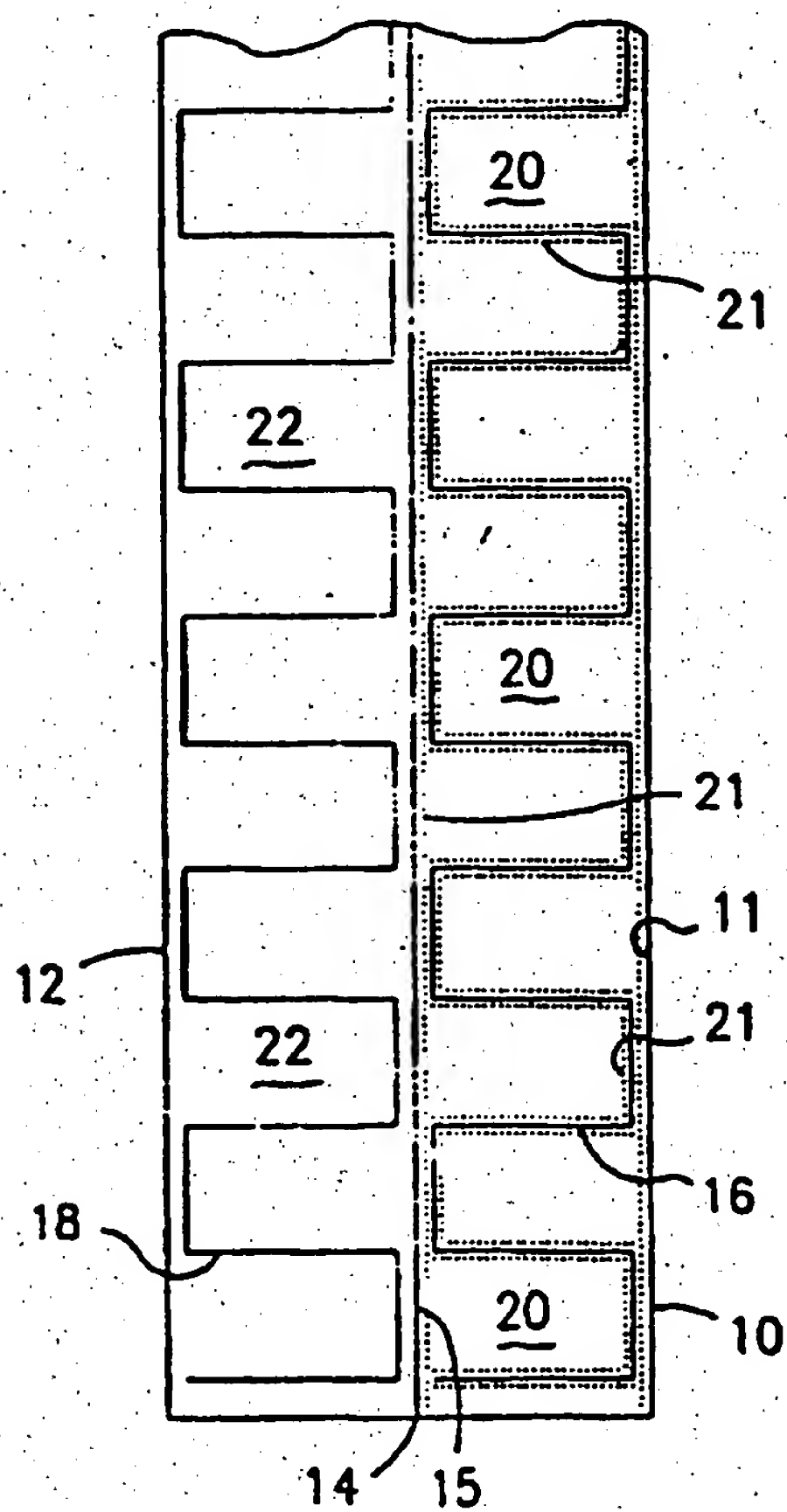


FIG. 3

INTERNATIONAL SEARCH REPORT

International Application No.
PCT/US 97/00491

A. CLASSIFICATION OF SUBJECT MATTER
 IPC 6 H01M8/06 B01J15/00 B01J12/00 B01J19/00 C01B3/58
 C10K3/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
 IPC 6 H01M B01J C01B C10K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	GB 1 604 980 A (MINNESOTA MINING & MFG) 16 December 1981 see page 3, line 22 - line 46; claims 1,4,14,15,17; figure 1 see page 2, line 49 - line 51	1,6,9
Y	---	2,3,5,7
Y	EP 0 650 923 A (DAIMLER BENZ AG) 3 May 1995 see column 1, line 45 - column 2, line 4 see column 2, line 32 - column 3, line 19 see column 4, line 1 - line 23; figures 2,3 ---	2
	-/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents:

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

Z document member of the same patent family

Date of the actual completion of the international search

6 June 1997

Date of mailing of the international search report

12. 06. 97

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
 NL - 2280 HV Rijswijk
 Tel. (+ 31-70) 340-2040, Tx. 31.651 epo nl,
 Fax (+ 31-70) 340-3016

Authorized officer

D'hondt, J

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 97/00491

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP 0 529 329 A (ISHIKAWAJIMA HARIMA HEAVY IND) 3 March 1993 see column 6, line 24 - column 7, line 12; figures 1-5 see column 8, line 31 - line 58 see column 5, line 6 - line 11 ---	3,5
X	WO 93 19005 A (INT FUEL CELLS CORP) 30 September 1993 see page 10, line 9 - line 25; claim 1; figure 1 see page 9, line 27 - line 31 ---	10
Y	---	7
A	PATENT ABSTRACTS OF JAPAN vol. 006, no. 054 (C-097), 9 April 1982 & JP 56 166937 A (OSAKA GAS CO LTD), 22 December 1981, see abstract ---	1
A	EP 0 618 003 A (SULZER CHEMTECH AG) 5 October 1994 see claims 1,11; figures 1-4 ---	1
A	WO 95 01834 A (PACKINOX ;UOP INC (US)) 19 January 1995 see claim 12; figures 1-3 see page 4, last paragraph - page 5, paragraph 1 see page 9, last paragraph see page 11, paragraph 3 ---	1
A	EP 0 305 203 A (TOPSOE HALDOR AS) 1 March 1989 see page 5, line 24 - line 25; claim 1; figure 3 ---	
P,X	EP 0 743 694 A (TOYOTA MOTOR CO LTD) 20 November 1996 see column 26, line 34 - line 48; claim 1; figure 14 see column 30, line 29 - line 42; figure 17 ---	10
E	DE 195 39 648 A (DAIMLER BENZ AG) 7 May 1997 see column 2, line 36 - column 4, line 19; claims 1,7,11; figures 1,2 see column 6, line 17 - line 25 -----	1-5,8,10

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No.

PCT/US 97/00491

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
GB 1604980 A	16-12-81	AU 522110 B AU 3743478 A BE 868451 A BR 7804053 A CA 1098288 A DE 2827934 A FR 2405743 A JP 54011077 A NL 7806779 A	20-05-82 03-01-80 27-12-78 09-01-79 31-03-81 11-01-79 11-05-79 26-01-79 29-12-78
EP 0650923 A	03-05-95	DE 4334981 A	20-04-95
EP 0529329 A	03-03-93	JP 5047402 A JP 5043201 A DE 69218518 D US 5270127 A	26-02-93 23-02-93 30-04-97 14-12-93
WO 9319005 A	30-09-93	EP 0631564 A US 5330727 A	04-01-95 19-07-94
EP 0618003 A	05-10-94	JP 6319989 A US 5467817 A	22-11-94 21-11-95
WO 9501834 A	19-01-95	FR 2707186 A FR 2708871 A FR 2708872 A CA 2166564 A CN 1128506 A EP 0707517 A EP 0766999 A EP 0767000 A US 5600053 A	13-01-95 17-02-95 17-02-95 19-01-95 07-08-96 24-04-96 09-04-97 09-04-97 04-02-97
EP 0305203 A	01-03-89	CA 1304215 A DE 3885545 D DE 3885545 T JP 1139137 A JP 2592662 B US 4985230 A	30-06-92 16-12-93 10-03-94 31-05-89 19-03-97 15-01-91

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 97/00491

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0743694 A	20-11-96	JP 9030802 A	04-02-97
DE 19539648 A	07-05-97	NONE	

INTERNATIONAL SEARCH REPORT

THIS PAGE BLANK (USPTO)